



VICTORIA GOVERNMENT GAZETTE

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No. 3.]

FRIDAY, JANUARY 10.

[1908.]

Mining Notices.

ELECTRO-HYDRAULIC DREDGING COMPANY NO LIABILITY.

AN Extraordinary Meeting of the Shareholders of the above-named company will be held at the company's Office, National Mutual Buildings, Ballarat, at the hour of a quarter-past Two p.m., on the 24th day of January, 1908:—

Business:

1. To authorize the directors of the said company to borrow money on the security of a mortgage or bill of sale of the property of the said company, or any part thereof, and for that purpose to affix the company's seal to such mortgage or bill of sale.

2. To confirm the minutes of the meeting.

Dated this 9th day of January, 1908.

966 WM. LASCELLES, Manager.

OLD JUBILEE GOLD MINES NO LIABILITY.
OWING to the 10th Call of Threepence per share on the uncalled capital of the company having been illegally made, the same is hereby cancelled.

By order of the Board,

971 W. A. RENOU, Acting Manager.

WELSH AND SLATER GOLD MINING COMPANY NO LIABILITY.

ALL shares forfeited for non-payment of the 6th call of One penny per share will be sold by public auction, at the Stock Exchange, Collins-street, Melbourne, on Saturday, 18th January, 1908, at a quarter-past Eleven a.m., unless such call be previously paid.

By order of the Board,

967 TOM HILLS
(Hills and Melvin), Manager.

MOUNT TARRANGOWER GOLD MINING COMPANY NO LIABILITY.

NOTICE is hereby given that all shares forfeited for the non-payment of the 72nd (October) call of One penny per share will be sold by public auction, at the Stock Exchange, Melbourne, on Saturday, the 18th day of January, 1908, at half-past Eleven o'clock a.m., unless previously redeemed.

By order of the Board,

968 W. A. BUTLER, Manager.

OLD WILLIAM TELL & LAST CHANCE GOLD MINING COMPANY NO LIABILITY.

ALL shares forfeited for non-payment of the 3rd call of Two-pence per share will be sold by public auction, at the registered office, 34 Queen-street, Melbourne, on Tuesday, 21st January, 1908, at a quarter to Three o'clock in the afternoon, unless previously redeemed.

969 ARTHUR PEARSON, Manager.

SOUTH VICTORIAN CORNISH GOLD MINING COMPANY NO LIABILITY.

ALL shares forfeited for non-payment of the 10th call of Twopence per share will be sold by public auction, at the Stock Exchange, Melbourne, on Tuesday, 21st January, 1908, at Three o'clock in the afternoon, unless previously redeemed.

ARTHUR PEARSON, Manager.

Registered office: 34 Queen-street, Melbourne. 970

MOUNT STRONACH TIN MINES NO LIABILITY.
NOTICE is hereby given that all shares upon which the 8th call of Three halfpence per share (and any previous calls) remain unpaid will be sold by public auction, at the Vestibule of the Stock Exchange, Collins street, Melbourne, on Saturday, 18th January, 1908, at half-past Eleven o'clock a.m.

By order of the Board,

70 Elizabeth-street, Melbourne, 8th January, 1908. 972
T. P. HUSBAND, Manager.

SPRING GULLY JUNCTION GOLD MINING COMPANY NO LIABILITY.

NOTICE.—All shares on which the 28th call of Threepence per share remains unpaid are forfeited, and will be sold by W. P. Bentley and Co., at Beehive Exchange, Bendigo, on Saturday, 18th January, 1908, at half-past Four o'clock p.m., unless previously paid on.

973 G. A. PETRIE, Manager.

MORNING STAR OPTION COMPANY NO LIABILITY.

NOTICE.—All shares on which the 32nd call of Threepence per share remains unpaid are forfeited, and will be sold by Thos. Morrow and Co., at Beehive Exchange, Bendigo, on Saturday, 18th January, 1908, at half-past Four o'clock p.m., unless previously paid on.

974 G. A. PETRIE, Manager.

By Authority: J. KEMP, Acting Government Printer, Melbourne.

7-11-19

1. The first part of the report is a general description of the project. It includes the title, the objectives, the scope, and the methodology. The title is "The Effect of Temperature on the Rate of Reaction of Hydrogen Peroxide with Potassium Iodide". The objectives are to determine the effect of temperature on the rate of reaction and to determine the activation energy of the reaction. The scope is to study the reaction of hydrogen peroxide with potassium iodide at different temperatures. The methodology is to use a colorimetric method to measure the rate of reaction.

2. The second part of the report is a description of the experimental procedure. It includes the list of materials, the list of equipment, and the step-by-step procedure. The materials are hydrogen peroxide, potassium iodide, and sulfuric acid. The equipment is a water bath, a thermometer, a stopwatch, and a colorimeter. The procedure is to prepare a series of solutions of different concentrations of hydrogen peroxide and potassium iodide, to measure the rate of reaction at different temperatures, and to plot the rate of reaction against temperature.

3. The third part of the report is a description of the results. It includes the data table, the graphs, and the conclusions. The data table shows the rate of reaction at different temperatures. The graphs show the rate of reaction against temperature and the activation energy of the reaction. The conclusions are that the rate of reaction increases with temperature and that the activation energy of the reaction is 50 kJ/mol.

4. The fourth part of the report is a discussion of the results. It includes the interpretation of the results, the evaluation of the experiment, and the suggestions for further work. The interpretation of the results is that the rate of reaction increases with temperature because the molecules have more energy and are more likely to collide successfully. The evaluation of the experiment is that the results are in good agreement with the theoretical predictions. The suggestions for further work are to study the effect of other factors on the rate of reaction and to determine the order of the reaction.

5. The fifth part of the report is a conclusion. It summarizes the main findings of the experiment and states the overall conclusion. The main findings are that the rate of reaction increases with temperature and that the activation energy of the reaction is 50 kJ/mol. The overall conclusion is that the experiment was successful in determining the effect of temperature on the rate of reaction and in determining the activation energy of the reaction.